

Current status of a 1Ah class solid-state lithium metal secondary battery: Cycle and Rate capability

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Background

2

Pelletized cell: intrinsic characteristics

3

Prototype cell: current status

4

Conclusions

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Background

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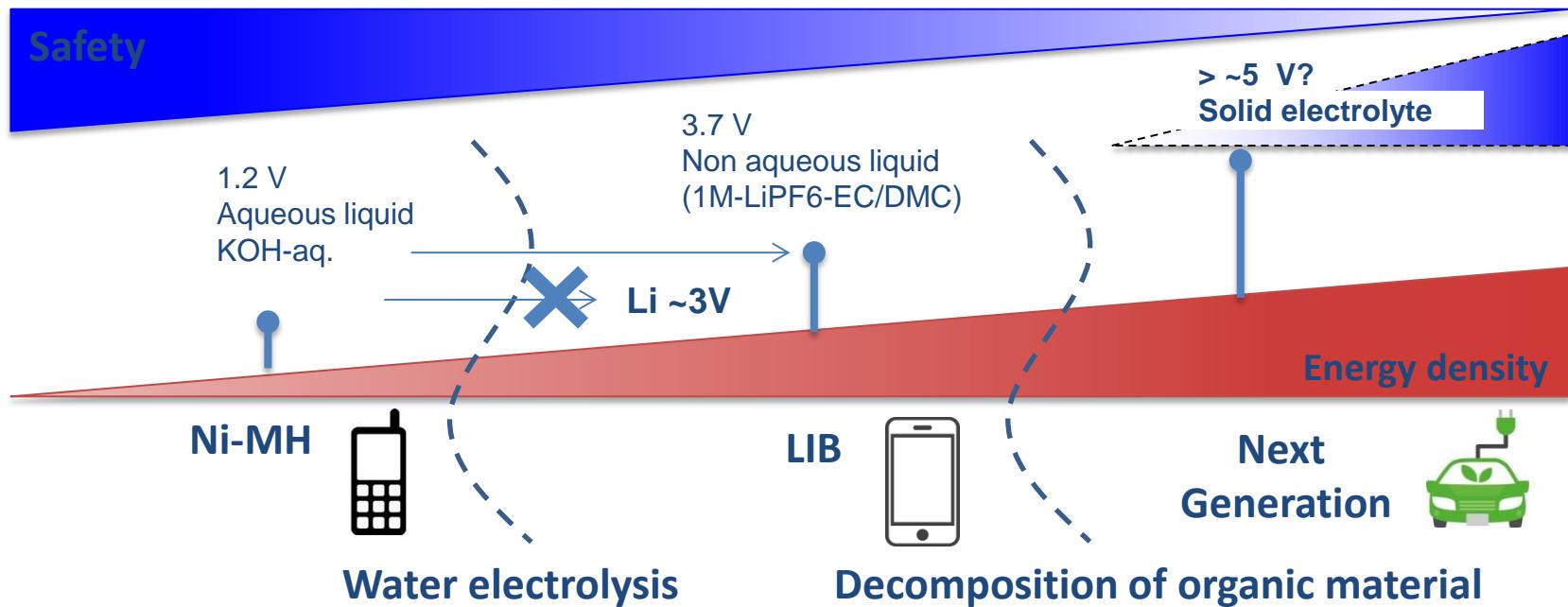
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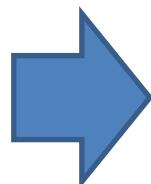
Conclusions



- Desired properties -

- ✓ **Electrochemical stability**
- ✓ **Ideal Li⁺ transport and reaction**
- ✓ **Safety (Strategy for Li-metal anode)**

1. Softness of sulfide based electrolyte
2. High ionic conductivity $>2 \text{ mScm}^{-1}$, $t^+=1$
3. Buffer layer on cathode active material



Large format cell

Issues:

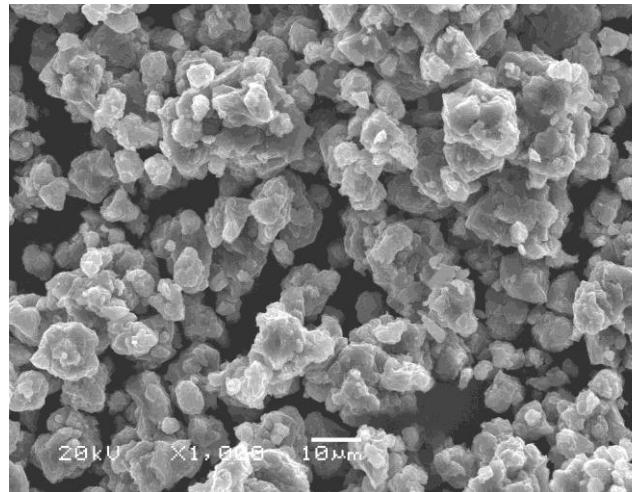
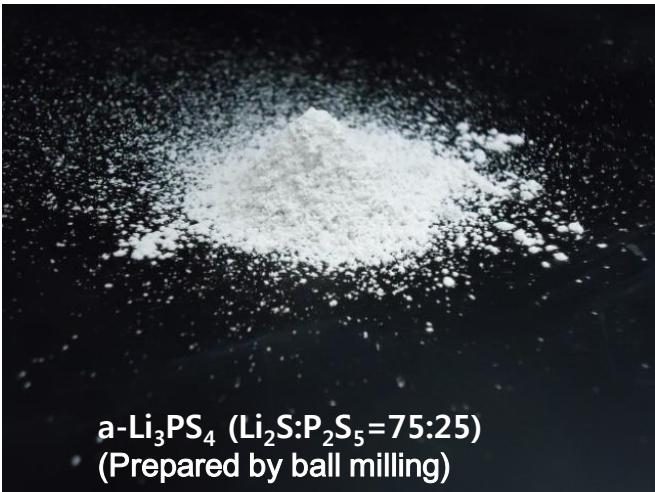
- Process (compression/tact time)
- Supply chain
- Operation pressure
- Hydrolysis of sulfide materials



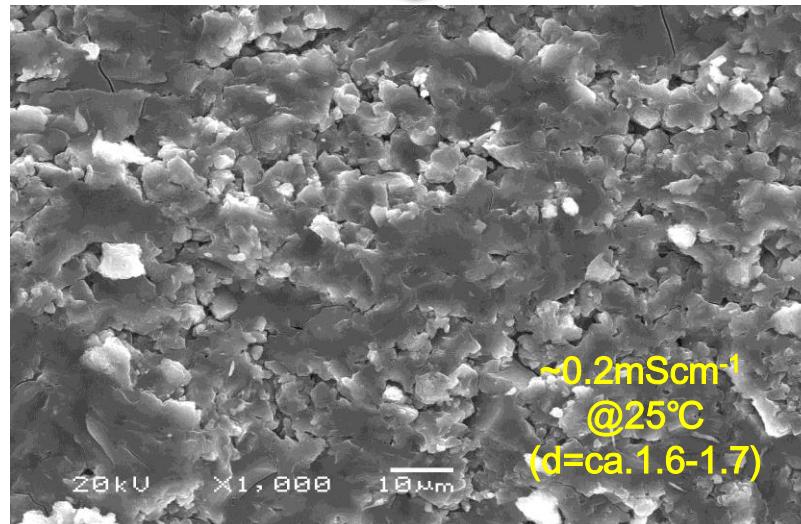
resolution

Mass Production

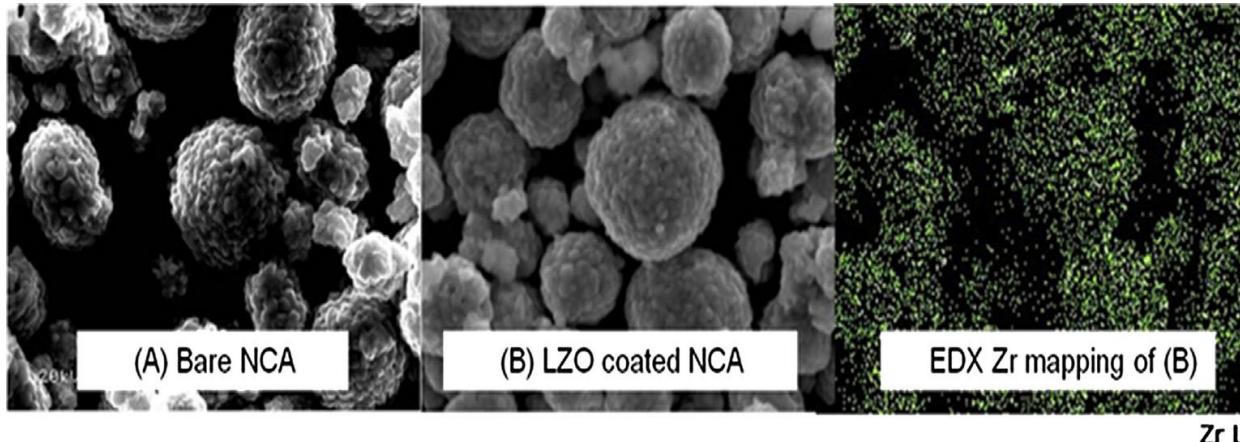
Softness of sulfide SEs



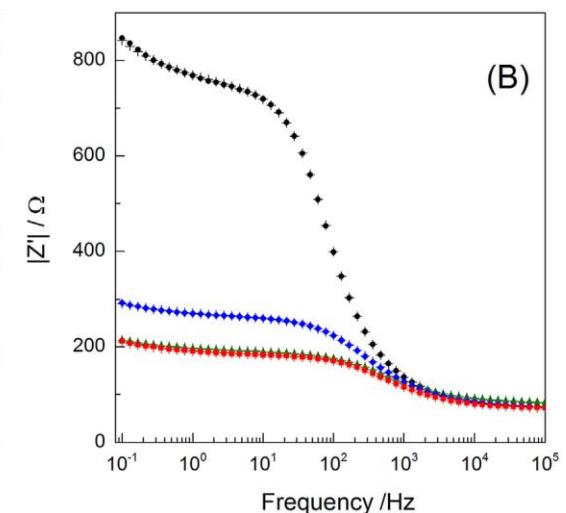
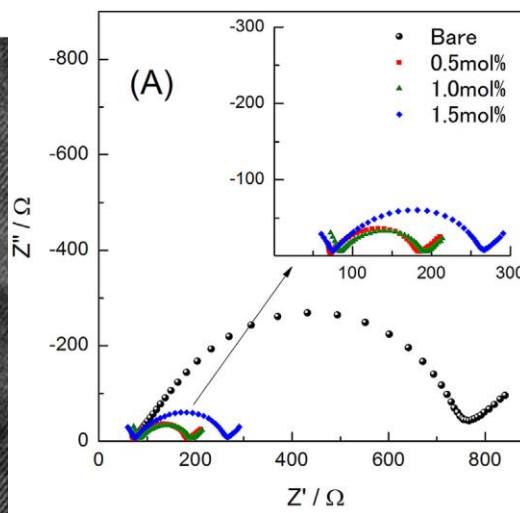
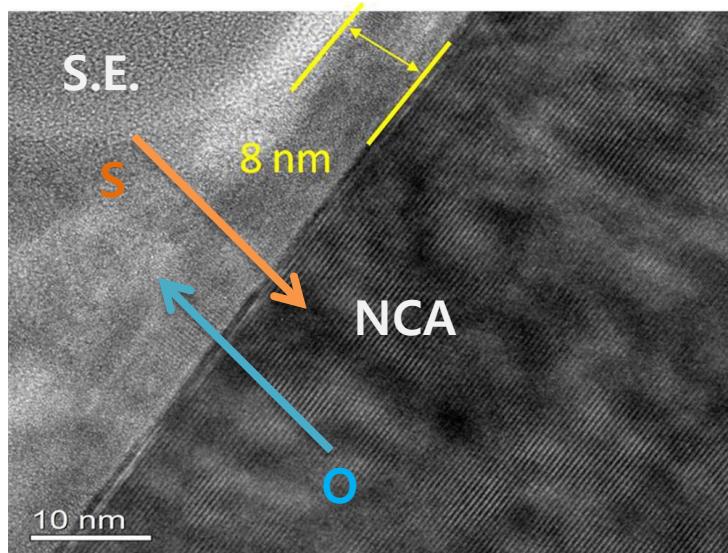
Cold press



➤ Mutual diffusion must be prevented.

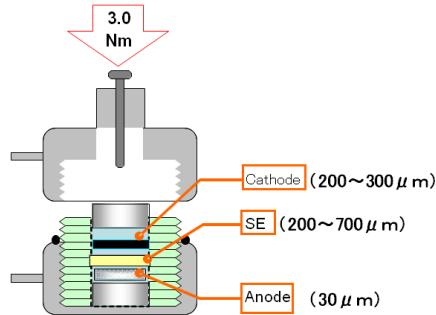


$R_{CT} \uparrow$ at high SOC



Ito et al., Journal of Power Source 248 (2014)943-950

R&D time line at Samsung

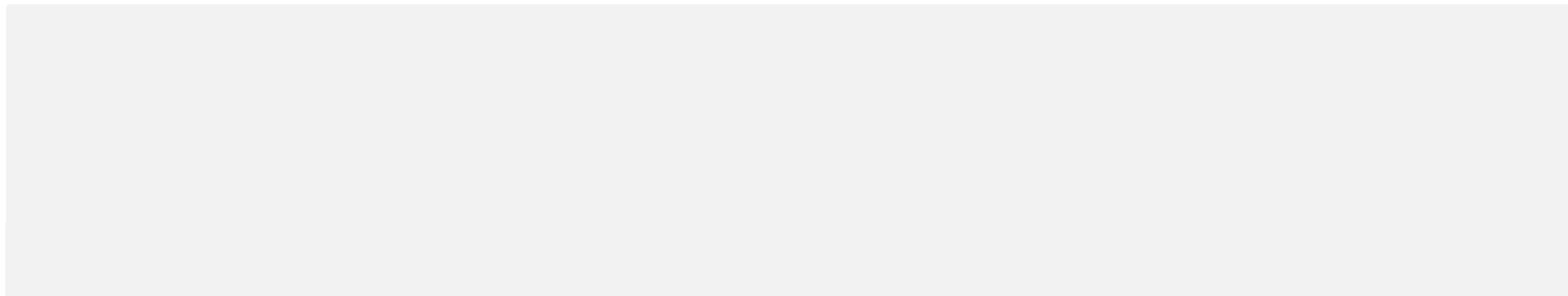


~2010

2014

2018

σ_{25}
P_(S.C.)
N_(S.C.)
Cell Q



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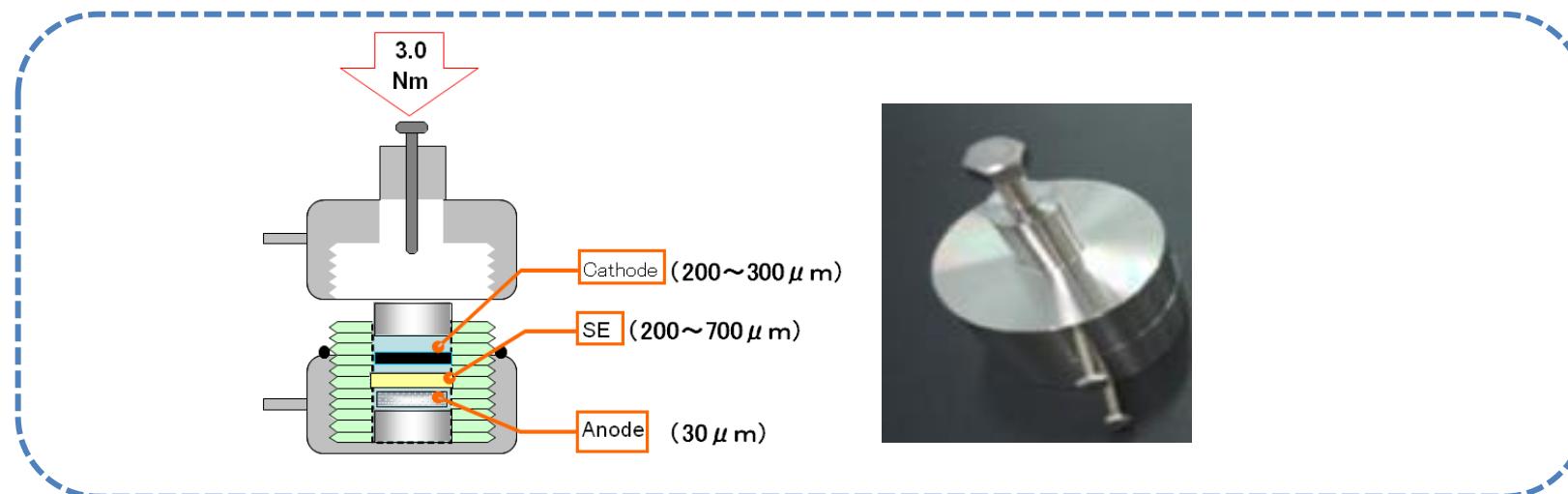
Conclusions

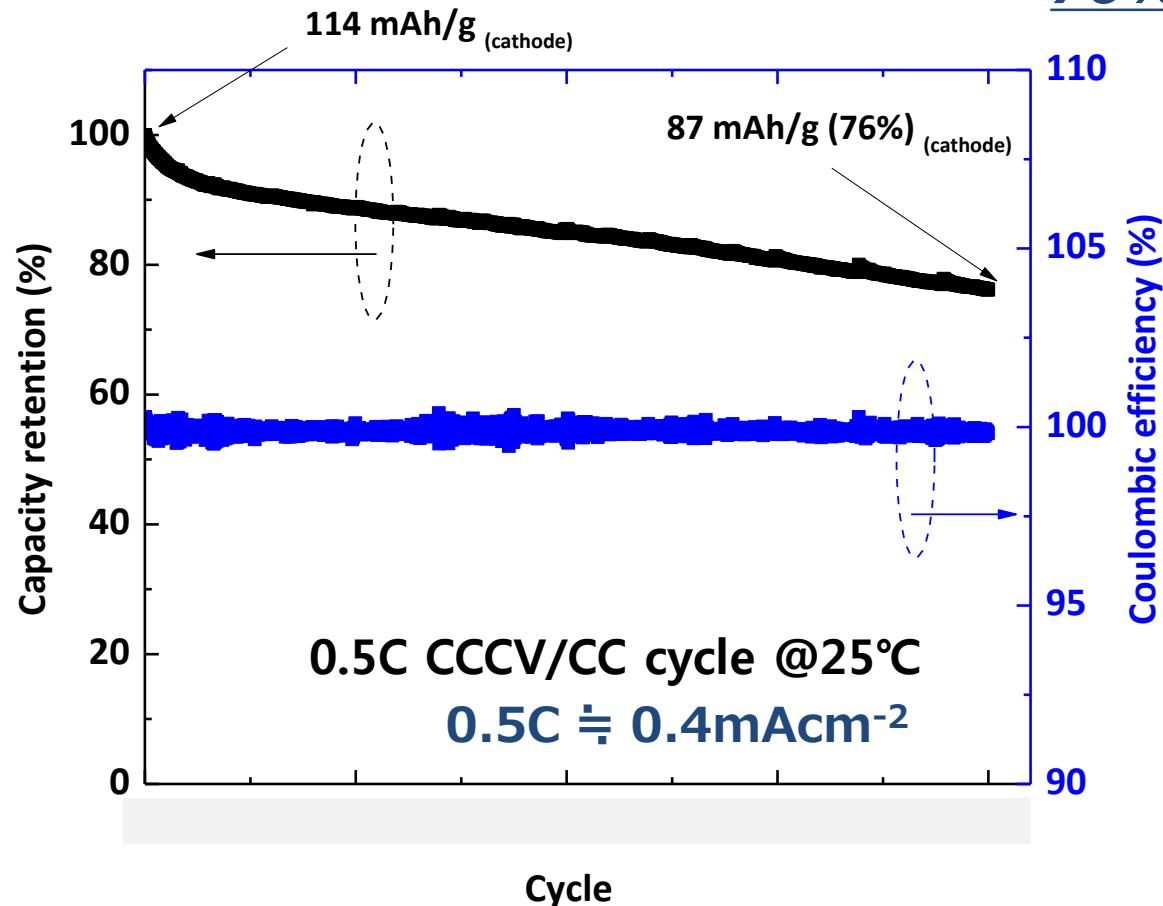
Preliminary test of a pelletized cell

- Test cell for materials evaluation. (Neglect energy density.)
- No binder → Ideal performance



- 0.8~1.5 mAh
- Graphite, Si, Li/SE/LMeO, S
- Characterization of materials
- No binder (pellet)
- 13 mmΦ (1.33cm²)





Item	Spec.
Negative	Li foil t=30μm
Positive (wt%)	NCA:SE:C=60:35:5
Loading	0.83 mAh/cm ²
Cap. Ratio N/P	26
SE	LPSX
Cell capacity	1.1mAh (cathode)

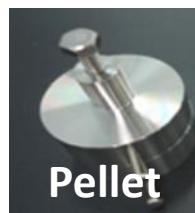
- Over 99.9% coulombic efficiency
- Negligible impact of reaction products

NCM: $\text{LiNi}_{0.8}\text{Co}_{0.1}\text{Mn}_{0.1}\text{O}_2$

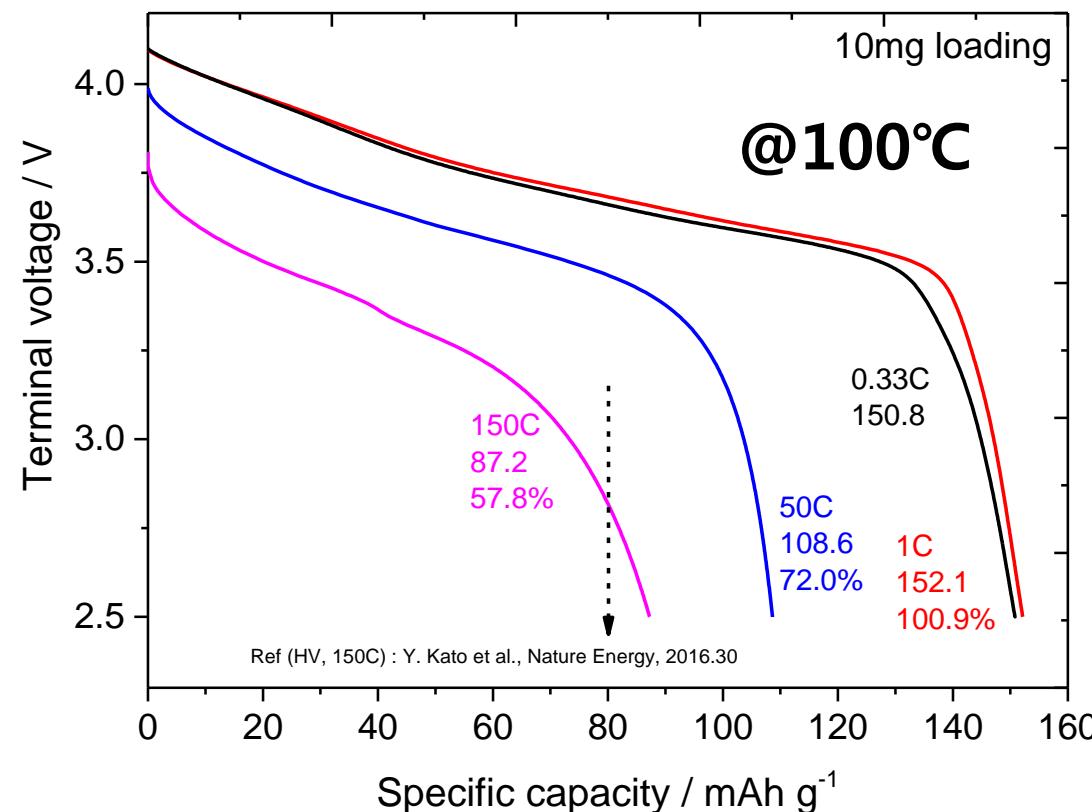
High rate discharge capability of a solid-state lithium secondary cell

13mmΦ pellet cell

Item	Spec.
Negative	Li foil t=30μm
Positive (wt%)	NCM:SE:C=60:35:5
Loading	0.6 mAh/cm ²
Cap. Ratio N/P	21
SE	$\text{LiCl}-\text{Li}_2\text{S}-\text{Li}_3\text{PS}_4$
Cell capacity	0.8 mAh (cathode)



Pellet



➤ 58% discharge at 150C (24sec)
@100°C

A large gap : Pellet and Prototype cell

- Continuous, large scale processes require wet process and binders.
- Unlike conventional LiBs, binders in ASSBs are insulator.
→ Significant reduction in electronic/ionic conductivity



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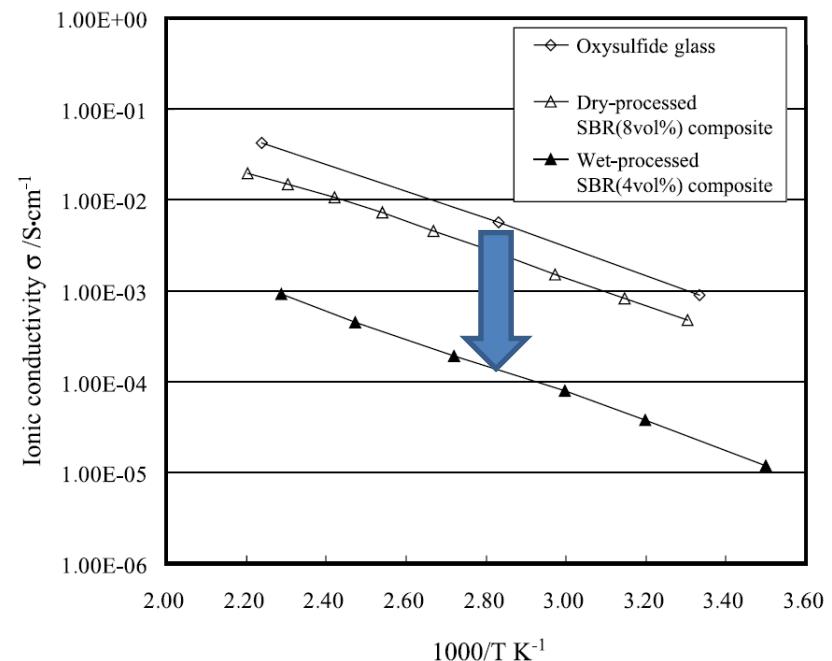
Solid State Ionics 158 (2003) 275–280

**SOLID
STATE
IONICS**
www.elsevier.com/locate/ssi

Fabrications and properties of composite solid-state electrolytes

Taro Inada^{a,*}, Kazunori Takada^a, Akihisa Kajiyama^a, Masaru Kouuchi^a,
Hideki Sasaki^a, Shigeo Kondo^a, Mamoru Watanabe^a,
Masahiro Murayama^b, Ryoji Kanno^b

^a Advanced Materials Laboratory, National Institute for Materials Science, 1-1 Namiki, Tsukuba, Ibaraki, 305-0044 Japan
^b Department of Electronic Chemistry, Interdisciplinary Graduate School of Science and Engineering, Tokyo Institute of Technology,
4259 Nagatsuta-cho, Midori-ku, Yokohama 226-8502, Japan



- Takada et al. propose dry process

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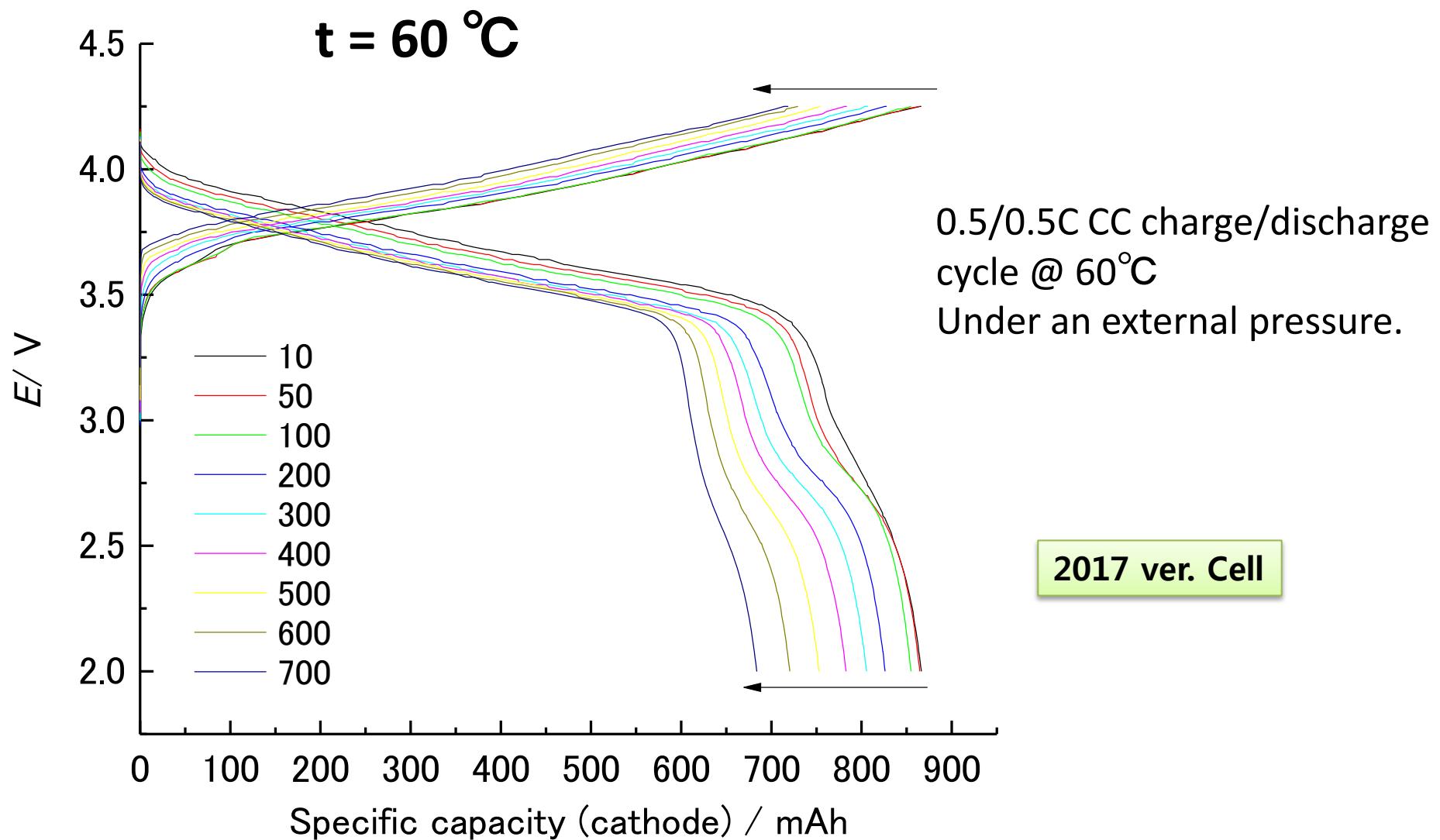
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1: Sulfide solid electrolyte

- Applicable (Stable operation at Li potential. High voltage is more challenging.)
- 700 cycles, demonstrated 60°C operation

2: Adaptation of Li-metal anode

- Excellent C.E. : No excess Li required
- Remaining issues :
 - i) Need external pressure
 - ii) Micro short circuit

3: Deposition-type Secondary ASSB

Very promising, but many outstanding issues
and need to ensure safety

- Organization committees
- Samsung Advanced Institute Technology
- SRJ/SAIT research staff